

1. Department Wise Information Regarding Faculty

Faculty Name	Designation	Pay scale with AGP	Email
Dr. Anu Sharma	Assistant Professor	15,600/- 39,100/- +6000/- AGP	issaranu@gmail.com

2. Format for Syllabus Break up

I. Subject: Engineering Chemistry Lab

Subject Code: 107

Class: B.Tech I Year (I Sem.)

Branch: CS-1(A1,A2,A3)

Branch: EE (F1, F2, F3)

Topic	practical classes required to cover the Topic	Month in which the topic will be covered	Actual date of covering the Topic
First Week (6th-8th Aug.,2015)			
Introduction: Laboratory Safety Measures	01	August 2015	F2, August,6 th (Thurs)
Introduction: Laboratory Safety Measures	01	August 2015	F3, August,7 th (Fri)
Introduction: Laboratory Safety Measures	01	August 2015	F1, August,8 th (Sat)
Second Week(10th-15th Aug.,2015)			
Introduction: Laboratory Safety Measures	1	August 2015	A2, August,10 th (Mon)
Introduction: Laboratory Safety Measures	1	August 2015	A3, August,11 th (Tue)
Introduction: Laboratory Safety Measures	1	August 2015	A1, August,12 th (Wed)
Basics of Practicals	1	August 2015	F2, August,13 th (Thurs)
Basics of Practicals	1	August 2015	F3, August14 th (Fri)
Third Week(17th-22nd Aug.,2015)			
Introduction: Basics of Practicals	1	August 2015	A2, August,17 th (Mon)
Introduction: Basics of	1	August 2015	A3, August,18 th (Tue)

Practicals			
Introduction: Basics of Practicals	1	August 2015	A1, August,19 th (Wed)
Introduction: Basics of Analysis	1	August 2015	F2, August,20 th (Thurs)
Introduction: Basics of Analysis	1	August 2015	F3, August 21 th (Fri)
Introduction: Basics of Practicals ,Basics of Analysis	1	August 2015	F1, August,22 nd (Sat)
Fourth Week(24th-29th Aug.,2015)			
Introduction: Basics of Analysis	1	August 2015	A2, August,24 th (Mon)
Introduction: Basics of Analysis	1	August 2015	A3, August,25 th (Tue)
Introduction: Basics of Analysis	1	August 2015	A1, August,26 th (Wed)
Introduction: Types of Titration	1	August 2015	F2, August,27 th (Thurs)
Introduction: Types of Titration	1	August 2015	F3, August,24 th (Fri)
Fifth Week (31st Aug.,2015)			
Types of Titration, To determine the strength of Ferrous Ammonium sulphate solution with the help of K₂Cr₂O₇ solution	1	August ,2015	A2, August,31 st (Mon)
First Week (1st-5th Sep.,2015)			
Types of Titration, To determine the strength of Ferrous Ammonium sulphate solution with the help of K₂Cr₂O₇ solution	01	Sep., 2015	A3, September ,1 st (Tue)
Types of Titration, To determine the strength of Ferrous Ammonium sulphate solution with the help of K₂Cr₂O₇ solution	01	Sep., 2015	A1, September,2 nd (Wed)
To determine the strength of Ferrous Ammonium sulphate solution with the help of K₂Cr₂O₇ solution	01	Sep., 2015	F2, September,3 th (Thurs)
To determine the strength of Ferrous Ammonium sulphate solution with the help of K₂Cr₂O₇ solution	01	Sep., 2015	F3,September,4 th (Fri)

Second Week (7th-12th Sep.,2015)			
To determine the strength of CuSO₄ solution with the help of hypo solution	01	Sep., 2015	A2, September,7 th (Mon)
To determine the strength of CuSO₄ solution with the help of hypo solution	01	Sep., 2015	A3, September ,8 th (Tue)
To determine the strength of CuSO₄ solution with the help of hypo solution	01	Sep., 2015	A1, September,9 th (Wed)
To determine the strength of CuSO₄ solution with the help of hypo solution	01	Sep., 2015	F2, September,10 th (Thurs)
To determine the strength of CuSO₄ solution with the help of hypo solution	01	Sep., 2015	F3,September,11 th (Fri)
Types of Titration, To determine the strength of Ferrous Ammonium sulphate solution with the help of K₂Cr₂O₇ solution	01	Sep., 2015	F1, September ,12 ^h (Sat)
Third Week (14th-19th Sep.,2015)			
To determine the Calorific Value of Solid Fuel by using Bomb Calorimeter	01	Sep., 2015	A2, September,14 th (Mon)
To determine the Calorific Value of Solid Fuel by using Bomb Calorimeter	01	Sep., 2015	A3, September ,15 th (Tue)
To determine the Calorific Value of Solid Fuel by using Bomb Calorimeter	01	Sep., 2015	A1, September16 th (Wed)
To determine the Calorific Value of Solid Fuel by using Bomb Calorimeter	01	Sep., 2015	F2, September,17 th (Thurs)
To determine the Calorific Value of Solid Fuel by using Bomb Calorimeter	01	Sep., 2015	F3,September,18 th (Fri)
To determine the strength of CuSO₄ solution with the help of hypo solution	01	Sep., 2015	F1, September ,19 th (Sat)
Fourth Week (21st-26th Sep.,2015)			
Practical Test: Basic of Practicals	01	Sep., 2015	A2, September,21 st (Mon)

Practical Test: Basic of Practicals	01	Sep., 2015	A3, September ,22 th (Tue)
Practical Test: Basic of Practicals	01	Sep., 2015	F2, September,24 th (Thurs)
To determine the Calorific Value of Solid Fuel by using Bomb Calorimeter, Practical Test: Basic of Practicals	01	Sep., 2015	F1, September ,26 th (Sat)
Fifth Week (28st-30th Sep.,2015)			
Practical Test:To determine the strength of Ferrous Ammonium sulphate solution with the help of K₂Cr₂O₇ solution	01	Sep., 2015	A2, September,28 th (Mon)
Practical Test:To determine the strength of Ferrous Ammonium sulphate solution with the help of K₂Cr₂O₇ solution	01	Sep., 2015	A3, September ,29 th (Tue)
Ist Mid Term			
First Week (1st-3rd Oct.,2015)			
Ist Mid Term			
Ist Mid Term			
Second Week (5th-10th Oct.,2015)			
Proximate analysis of solid fuel	01	Oct., 2015	A2,Oct.,5 th (Mon)
Proximate analysis of solid fuel	01	Oct., 2015	A3, Oct. , 6 th (Tue)
Proximate analysis of solid fuel	01	Oct., 2015	A1, Oct.7 th (Wed)
Proximate analysis of solid fuel	01	Oct., 2015	F2, Oct.,8 th (Thurs)
Proximate analysis of solid fuel	01	Oct., 2015	F3,Oct.,9 th (Fri)
Proximate analysis of solid fuel	01	Oct.Oct., 2015	F1, Oct. ,10 th (Sat)
Third Week (12th-17th Oct.,2015)			
To determine the viscosity of a given lubricating oil by Redwood viscometer.	01	Oct., 2015	A2,Oct.,12 th (Mon)
To determine the viscosity of a given lubricating oil by Redwood viscometer.	01	Oct., 2015	A1, Oct.14 ^h (Wed)
To determine the viscosity of a given lubricating oil	01	Oct., 2015	F2, Oct.,15 th (Thurs)

by Redwood viscometer.			
To determine the viscosity of a given lubricating oil by Redwood viscometer.	01	Oct., 2015	F3,Oct.,16 th (Fri)
To determine the viscosity of a given lubricating oil by Redwood viscometer.	01	Oct., 2015	F1,Oct.,17 th (Sat.)
Fourth Week (19th-24th Oct.,2015)			
Practical Test:To determine the Calorific Value of Solid Fuel by using Bomb Calorimeter	01	Oct., 2015	A2,Oct.,19 th (Mon)
To determine the viscosity of a given lubricating oil by Redwood viscometer.	01	Oct., 2015	A3, Oct. , 20 th (Tue)
Fifth Week (26th-31th Oct.,2015)			
To determine the strength of NaOH and Na ₂ CO ₃ in a given alkali mixture.	01	Oct., 2015	A2,Oct.,26 th (Mon)
To determine the strength of NaOH and Na ₂ CO ₃ in a given alkali mixture.	01	Oct., 2015	A3, Oct. , 27 th (Tue)
To determine the strength of NaOH and Na ₂ CO ₃ in a given alkali mixture.	01	Oct., 2015	A1, Oct.28 th (Wed)
To determine the strength of NaOH and Na ₂ CO ₃ in a given alkali mixture.	01	Oct., 2015	F2, Oct.,29 th (Thurs)
To determine the strength of NaOH and Na ₂ CO ₃ in a given alkali mixture.	01	Oct., 2015	F3,Oct.,30 th (Fri)
To determine the strength of NaOH and Na ₂ CO ₃ in a given alkali mixture.	01	Oct., 2015	F1,Oct.,31 st (Sat.)
First Week (02nd-07th Nov.,2015)			
Practical Test:To determine the strength of NaOH and Na₂CO₃ in a given alkali mixture.	01	Nov., 2015	A2,Nov.,02 nd (Mon)
Practical Test:To determine the strength of NaOH and Na₂CO₃ in a given alkali mixture.	01	Nov., 2015	A3, Nov., 03 rd (Tue)
Practical Test:To determine the strength of NaOH and Na₂CO₃ in a given alkali mixture.	01	Nov., 2015	A1, Nov.04 th (Wed)
Practical Test:To	01	Nov., 2015	F2,Nov,05 th (Thurs)

determine the strength of NaOH and Na₂CO₃ in a given alkali mixture.			
Practical Test:To determine the strength of NaOH and Na₂CO₃ in a given alkali mixture.	01	Nov., 2015	F3,Nov.,06 th (Fri)
Practical Test:To determine the strength of NaOH and Na₂CO₃ in a given alkali mixture.	01	Nov., 2015	F1,Nov.,07 st (Sat.)
Second Week (09nd -14th Nov.,2015)			
Practical Test:To determine the strength of CuSO₄ solution with the help of hypo solution	01	Nov., 2015	A2,Nov.,09 th (Mon)
Practical Test:To determine the strength of CuSO₄ solution with the help of hypo solution	01	Nov., 2015	A3, Nov., 10 th (Tue)
Third Week (16th -21st Nov.,2015)			
To determine the viscosity of a given lubricating oil by Redwood viscometer No. 1.	1	Nov., 2015	A2,Nov.,16 th (Mon)
To determine the viscosity of a given lubricating oil by Redwood viscometer No. 1.	1	Nov., 2015	A3, Nov., 17 th (Tue)
To determine the viscosity of a given lubricating oil by Redwood viscometer No. 1.	1	Nov., 2015	A1, Nov.18 th (Wed)
To determine the viscosity of a given lubricating oil by Redwood viscometer No. 1.	1	Nov., 2015	F2,Nov,19 th (Thurs)
To determine the viscosity of a given lubricating oil by Redwood viscometer No. 1.	1	Nov., 2015	F3,Nov.,20 th (Fri)
To determine the viscosity of a given lubricating oil by Redwood viscometer No. 1.	1	Nov., 2015	F1,Nov.,21 st (Sat.)
Fourth Week (24th -28th Nov.,2015)			
To determine cloud and pour point of a given oil.	1	Nov., 2015	A2,Nov.,24 th (Mon)

To determine cloud and pour point of a given oil.	1	Nov., 2015	A3, Nov., 25 th (Tue)
To determine cloud and pour point of a given oil.	1	Nov., 2015	A1, Nov.26 th (Wed)
To determine cloud and pour point of a given oil.	1	Nov., 2015	F2,Nov,27 th (Thurs)
To determine cloud and pour point of a given oil.	1	Nov., 2015	F3,Nov.,28 th (Fri)
To determine cloud and pour point of a given oil., and To Determine Flash & Fire Point of a given oil	1	Nov., 2015	F1,Nov.,29 th (Sat.)
Fifth Week (30th Nov.,2015)			
To determine the flash and fire point of a given lubricating oil.	1	Nov., 2015	A2,Nov.,30 th (Mon)
First Week (1st -5th Dec.,2015)			
To determine the flash and fire point of a given lubricating oil.	01	Dec., 2015	A3,Dec.,01 st (Tue)
To determine the flash and fire point of a given lubricating oil.	01	Dec., 2015	A1, Dec., 02 nd (Wed)
To determine the flash and fire point of a given lubricating oil.	01	Dec., 2015	F2, Dec.,03 rd (Thurs)
To determine the flash and fire point of a given lubricating oil.	01	Dec., 2015	F3,Dec.,04 th (Fri)
Internal Practical Exam,2015	1	Dec., 2015	F1,Dec.,05 th (Sat.)
Second Week (7th -11th Dec.,2015)			
Internal Practical Exam,2015	01	Dec., 2015	A2,Dec.,07 th (Mon)
Internal Practical Exam,2015	01	Dec., 2015	A3, Dec., 08 th (Tue)
Internal Practical Exam,2015	01	Dec., 2015	A1, Dec.,09 th (Wed)
Internal Practical Exam,2015	01	Dec., 2015	F2,Dec.,10 th (Thurs)
Internal Practical Exam,2015	01	Dec., 2015	F3,Dec.,11 th (Fri)
External Practical Exam,2015			

II A. 104 Engineering Chemistry
Format for Syllabus Break up of Theory
Branch: Computer Science Engineering- I

.No.	Topic	Lecture/ required to cover the topic	Month in which the topic will be covered	Actual date and day of covering the Topic CS1 (Tue, Thus., Fri.)
1.	UNIT-1 Interaction with Students, Brief of Syllabus and Mark distribution Pattern	1	August	Aug.06 th (Thurs.)
2.	Organic fuels, Origin, classification and general aspects of fossil fuels.	1	August	Aug.07 th (Fri.)
3.	Solid fuels, Coal, canonization of coal	1	August	Aug.11 th (Tue.)
4.	Manufacturing of coke by Beehive oven and by product oven method	1	August	Aug.13 th (Thurs.)
5.	Recovery of by products by Hoffman method, Liquid fuels, Composition of petroleum	1	August	Aug.14 th (Fri.)
6.	Advantages and refining of petroleum	1	August	Aug.18 th (Tue.)
7.	Cracking	1	August	Aug.20 th (Thurs.)
8.	Reforming, polymerization and isomerization of refinery products	1	August	Aug.21 th (Fri.)
9.	Synthetic petrol, Bergius and Fischer Tropsch process	1	August	Aug.25 th (Tue.)
10.	Knocking, octane number and anti- knocking agents.	1	Sep.	Aug.27 th (Thurs.)
11.	Gaseous fuels, Advantages, manufacturing	1	Sep.	Aug.28 th (Fri.)
12.	composition and calorific value of coal, gas and oil gas.	1	Sep.	Sep..01 th (Tue.)
13.	UNIT-2 Ultimate and proximate analysis of coal	1	Sep.	Sep.,03 th (Thurs.)
14.	Determination of calorific value of solid by bomb calorimeter	1	Sep.	Sep., .04 th (Fri.)

15.	Determination of calorific value of gaseous fuel by Junker's Calorimeter	1	Sep.	Sep.,08 th (Tue.)
16.	Calculations of calorific value based on Dulong's formula	1	Sep.	Sep.,10 th (Thurs.)
17.	Flue gas analysis by Orsat's apparatus and its significance.	1	Sep.	Sep., .11 th (Fri.)
18.	Numerical based on bomb Calorimeter	1	Sep.	Sep.,15 th (Tue.)
19.	Combustion, requirement of oxygen/air in combustion process.	1	Sep.	Sep.,17 th (Thurs.)
20.	Numerical based on bomb Calorimeter	1	Sep.	Sep., .18 th (Fri.)
21.	Combustion, requirement of oxygen/air in combustion process.	2	Sep.	Sep.,22 th (Tue.)
22.	UNIT-3 Polymers: Different methods of classification	1	Sep.	Sep.,24 th (Thurs.)
23.	Basic ideas of polymerization mechanisms	1	Sep.	Sep.,29 th (Tue.)
24.	Elastomers: Natural rubber, vulcanization,	1	Oct.	Oct.,01 (Thurs) I mid Term, Oct.,06 (Tue.)
25.	Synthetic Rubbers viz. Buna-S, Buna-N, Butyl and neoprene rubbers.	1	Oct.	Oct.,08 (Thurs.)
26.	Shown Evaluated Copies of I Mid Term	1	Oct.	Oct.,09 (Fri.)
27.	UNIT-3 Fullerenes: Introduction, properties, preparation and uses.	1	Oct.	Oct.,15 (Thurs.)
28.	Organic Electronic Materials (including conducting polymers- poly (p-phenylene)	1	Oct.	Oct.,16 (Fri.)
29.	polythiophenes, Polyphenylene, vinylenes, polypyroles, polyaniline).	1	Oct.	Oct.,20 (Tue.)

30.	UNIT-4 Cement: Definition, Composition, basic constituents and their significance	1	Oct.	Oct.,27 (Tue.)
31.	Manufacturing of Portland cement by Rotary Kiln Technology	1	Oct.	Oct.,29 (Thurs.),
32.	Chemistry of setting and hardening of cement and role of gypsum.	1	Nov.	Nov. 03,(Tue)
33.	UNIT-4 Glass: Definition, Properties, Manufacturing of glass	1	Nov.	Nov. 05, (Thus)
34.	Importance of annealing in glass making,Optical fiber grade glass	1	Nov.	Nov. 06,(Fri)
35.	Types of silicate glasses and their commercial uses	1	Nov.	Nov. 17,(Tue)
36.	UNIT-5 Refractory: Definition, classification, properties	1	Nov.	Nov. 19,(Thus)
37.	Requisites of good refractory and manufacturing of refractory.	1	Nov.	Nov. 20,(Fri)
38.	Preparation of Silica and fire clay refractory with their uses.		Nov.	Nov. 24,(Tue)
39.	Seger's (Pyrometric) Cone Test and RUL Test	1	Nov.	Nov. 26,(Thus)
40.	UNIT-5 Lubricants: Introduction, classification and uses of lubricants. Types of lubrication.	1	Nov.	Nov. 27,(Fri)
41.	Viscosity & viscosity index	1	Dec.	Dec.1,(Tue)
42.	Flash and fire point, cloud and pour point	1	Dec.	Dec.3,(Thus)
43.	Steam Emulsion Number,Precipitation number,neutralization Number.	1	Dec.	Dec.4,(Fri)
44.	Revision :Combustion Numerical/ Discuss Problem	1	Dec.	Dec.8,(Tue)
45.	Revision :Combustion Numerical/Discuss Problem	1	Dec.	Dec.10,(Thus.)
46.	Test Unit -3	1	Dec.	Dec.11,(Fri.)

II B. 104 Engineering Chemistry
Format for Syllabus Break up of Theory
Branch:Electrical Engineering

S.No.	Topics	Lecture/ required to cover the topic	Month in which the topic will be covered	Actual date and day of covering the Topic EE (Wed.,Fri, Sat)
1.	UNIT-1 Interaction with Students,Brief of Syllabus and Mark distribution Pattern	1	August	Aug.07 th (Fri.)
2.	Organic fuels, Origin, classification and general aspects of fossil fuels.	1	August	Aug.08 th (Sat.)
3.	Solid fuels, Coal, canonization of coal	1	August	Aug.12 th (Wed.)
4.	Manufacturing of coke by Beehive oven and by product oven method	1	August	Aug.14 th (Fri.)
5.	Recovery of by products by Hoffman method, Liquid fuels, Composition of petroleum	1	August	Aug.19 ^h (Wed.)
6.	Advantages and refining of petroleum	1	August	Aug.21 th (Fri.)
7.	Cracking	1	August	Aug.22 th (Sat.)
8.	Reforming, polymerization and isomerization of refinery products	1	August	Aug.26 ^h (Wed.)
9.	Synthetic petrol, Bergius and Fischer Tropsch process	1	August	Aug.28 th (Fri.)
10.	Knocking, octane number and anti- knocking agents.	1	Sep.	Sep., 2 th (Wed.)
11.	Gaseous fuels, Advantages, manufacturing	1	Sep.	Sep., .04 th (Fri.)

12.	composition and calorific value of coal, gas and oil gas.	1	Sep.	Sep., 9 th (Wed.)
13.	UNIT-2 Ultimate and proximate analysis of coal	1	Sep.	Sep., .11 th (Fri.)
14.	Determination of calorific value of solid by bomb Calorimeter	1	Sep.	Sep., .12 th (Sat.)
15.	Determination of calorific value of gaseous fuel by Junker's Calorimeter	1	Sep.	Sep.,16 th (Wed.)
16.	Calculations of calorific value based on Dulong's formula	1	Sep.	Sep., .18 th (Fri.)
17.	Flue gas analysis by Orsat's apparatus and its significance.	1	Sep.	Sep., .19 th (Sat.)
18.	Numerical based on bomb Calorimeter	1	Sep.	Sep., .26 th (Sat.)
19.	Combustion, requirement of oxygen/air in combustion process.	1	Sep. /Oct.	Sep.,30th (Wed.)I mid Term,03 (Sat.),I mid Term, Oct.,07 (Wed.)
20.	Numerical based on bomb Calorimeter	1	Oct.	Oct.,09 (Fri.)
21.	Combustion, requirement of oxygen/air in combustion process.	2	Oct.	Oct.,10 (Sat.)
22.	UNIT-3 Polymers: Different methods of classification	1	Oct.	Oct.,14 (Wed.)
23.	Basic ideas of polymerization mechanisms	1	Oct.	Oct.,16 (Fri.)
24.	Elastomers: Natural rubber, vulcanization,	1	Oct.	Oct.,17 (Sat.)
25.	Synthetic Rubbers viz. Buna-S, Buna-N, Butyl and neoprene rubbers.	1	Oct.	Oct.,28 (Wed.)
26.	Shown Evaluated Copies of I Mid Term	1	Oct.	Nov. 04, (Wed)

27.	UNIT-3 Fullerenes: Introduction, properties, preparation and uses.	1	Oct.	Nov. 06,(Fri)
28.	Organic Electronic Materials (including conducting polymers- poly (p-phenylene)	1	Oct.	Nov. 07,(Sat)
29.	polythiophenes, Polyphenylene, vinylenes, polypyrroles, polyaniline).	1	Oct.	Nov. 18, (Wed)
30.	UNIT-4 Cement: Definition, Composition, basic constituents and their significance	1	Nov.	Nov. 20,(Fri)
31.	Manufacturing of Portland cement by Rotary Kiln Technology	1	Nov.	Nov. 21,(Sat)
32.	Chemistry of setting and hardening of cement and role of gypsum.	1	Nov.	Nov. 27,(Fri)
33.	UNIT-4 Glass: Definition, Properties, Manufacturing of glass,Importance of annealing in glass making,Optical fiber grade glass	1	Nov.	Nov. 28,(Sat)
34.	Types of silicate glasses and their commercial uses UNIT-5 Refractory: Definition, classification, properties,Requisites of good refractory and manufacturing of refractory,	1	Dec.	Dec. 2(Wed)
35.	Seeger's(Pyrometric) Cone Test and RUL Test Preparation of Silica and fire clay refractory with their uses.	1	Dec.	Dec. 4(Fri.)
36.	UNIT-5 Lubricants: Introduction, classification and uses of lubricants. Types of lubrication.,Viscosity & viscosity index	1	Dec.	Dec. 5(Sat.)
37.	Flash and fire point, cloud and pour point	1	Dec.	Dec. 9(Wed)

38.	Steam Emulsion Number, Precipitation number, neutralization Number.		Dec.	Dec. 11(Fri.)
II Mid Term 21/12/15 to 23/12/15				

**II B. 104 Engineering Chemistry
Format for Syllabus Break up of Tutorial
Branch: Electrical Engineering**

S.No.	Topic	Tutorial/ required to solve the topic problems	Month & Date in which the topic will be covered EE(F1)
1.	Definition and classifications Fuels	1	Aug. 11 th ,
2.	Coke Production and Synthetic Fuels	1	Aug., 18 th
3.	Carbonisation of Fuels and refining of Petroleum	1	Aug., 25 th
4.	Practice on Ultimate and proximate analysis of coal Numerical, Dulong Numerical	1	Sep. 1 st
5.	Practice on Bomb numerical, Junker's based Numerical	1	Sep., 08 th
6.	Combustion (By Weight) based numerical	1	Sep., 15 th
7.	Combustion (By Weight, By Volume) based numerical,	1	Sep., 22 nd
8.	Combustion (By volume) based numerical	1	Sep., 29 th
9.	Polymers , Vulcanization	1	Oct. 06 th
10.	New Engineering Material	1	Oct., 1 st
11.	Test: Unit- I, II	1	Oct., 20 th
12.	Test: Unit- II	1	Oct., 27 th

13.	Cement,Glass	1	Nov., 03 rd
14.	Refractory	1	Nov.,17 th
15.	Types of Refractory	1	Nov.,24 th
16.	Lubrication	1	Dec. 01 st
17.	Properties of Lubricants	1	Dec. ,08 th

3. Time Table of the Faculty (Dr. Anu Sharma)

104 Engineering Chemistry

B. Tech. I Semester Session 2015-16

Day/Time	8.00-8.50	8.50-9.40	9.40-10.30	10.30 -11.20	11.20-12.10	12.10-1.00
Mon	A2 Lab.					
Tues		F1(Tut.)	A3 Lab.			CS-I (Lec.)
Wed			A1 Lab.		EE (Lec.)	
Thurs	F2 Lab.			CS-I (Lec.)		
Fri	F3 Lab.		CS-I (Lec.)		EE (Lec.)	
Sat		EE (Lec.)	F1 Lab.			

4. Time Table of Engineering Chemistry Lab

Subject Code: 107

B. Tech. I Semester Session 2015-16

Time (1.40 hr)/ day	8.00- 9.40 am	9.40-11.20 am	11.20-1.00 am
Monday	A2 Chem. Lab (AS)	B2 Chem. Lab (KS)	C2 Chem. Lab (GS)
Tuesday	B3 Chem. Lab (KS)	A3 Chem. Lab (AS)	C3 Chem. Lab (GS)

Wednesday	B1 Chem. Lab (KS)	A1 Chem. Lab (AS)	C1 Chem. Lab (GS)
Thursday	F2 Chem. Lab (AS)	E1 Chem. Lab (GS)	D2 Chem. Lab (SKM)
Friday	F3 Chem. Lab (AS)	E2 Chem. Lab (GS)	D1 Chem. Lab (SKM)
saturday	E3 Chem. Lab (GS)	F1 Chem. Lab (AS)	D3 Chem. Lab (SKM)

AS: Dr. Anu Sharma ,
KS: Dr. Kanika Solanki

SKM: Dr. S.K.Mehla

GS: Dr. Gayatri Sharma